

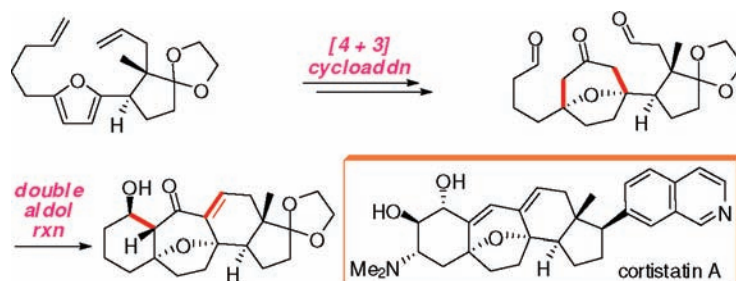
Concise Synthesis of the
Oxapentacyclic Core of Cortistatin AFangmiao Yu,[†] Guozhi Li,[†] Peng Gao,[†] Hongju Gong,[†] Yinghua Liu,[†]
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ABSTRACT



A concise synthetic approach for constructing the oxapentacyclic framework of cortistatin A is described. The synthesis features a furan–oxallyl [4 + 3] cycloaddition and double-intramolecular aldol reactions. In addition, an interesting core structure was obtained in 11 steps from furan by using our method.

Cortistatins, containing a common unprecedented [6.7.6.5] oxapentacyclic [or termed as 9-(10,19)-abeo-androstane] skeleton, were isolated from the marine sponge *Corticium simplex* by Kobayashi and co-workers and were found to possess potent antiangiogenic and antiproliferative activities against human umbilical vein endothelial cells (HUVECs).¹ Among these natural substances, cortistatin A (**1**, Figure 1a) demonstrated the strongest antiproliferative activity (IC₅₀ = 1.8 nM) against HUVECs. It also had a selectivity index that was 3000 times different from that of normal fibroblasts and various tumor cell lines. The unique structural charac-

teristics and remarkable pharmacological profiles of cortistatins have spurred synthetic chemists to develop efficacious and practical synthetic routes to them for further biological investigations.² Until now, elegant semi-, total, and formal syntheses of **1** have been accomplished by the teams of Baran,³ Nicolaou,⁴ Shair,⁵ Hirma,⁶ Sarpong,⁷ and Myers,⁸ respectively, and synthetic studies toward the construction

(2) For reviews on the synthesis of cortistatins, see: (a) Hardin, A. R.; Simmons, E. M.; Sarpong, R. *Eur. J. Org. Chem.* **2010**, 3553. (b) Shi, Y.; Tian, W. *Chin. J. Org. Chem.* **2010**, *30*, 515.

(3) Shenvi, R. A.; Guerrero, C. A.; Shi, J.; Li, C.-C.; Baran, P. S. *J. Am. Chem. Soc.* **2008**, *130*, 7241 (semi-synthesis).

(4) (a) Nicolaou, K. C.; Sun, Y.-P.; Peng, X.-S.; Polet, D.; Chen, D. Y.-K. *Angew. Chem., Int. Ed.* **2008**, *47*, 7310 (total synthesis). (b) Nicolaou, K. C.; Peng, X.-S.; Sun, Y.-P.; Polet, D.; Zou, B.; Lim, C. S.; Chen, D. Y. K. *J. Am. Chem. Soc.* **2009**, *131*, 10587 (total synthesis).

(5) Lee, H. M.; Nieto-Oberhuber, C.; Shair, M. D. *J. Am. Chem. Soc.* **2008**, *130*, 16864 (total synthesis).

(6) Yamashita, S.; Kitajima, K.; Iso, K.; Hirma, M. *Tetrahedron Lett.* **2008**, *49*, 6613 (formal synthesis).

(7) Simmons, E. M.; Hardin-Narayan, A. R.; Guo, X.; Sarpong, R. *Tetrahedron* **2010**, *66*, 4696 (formal synthesis).

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(1) (a) Aoki, S.; Watanabe, Y.; Sanagawa, M.; Setiawan, A.; Kotoku, N.; Kobayashi, M. *J. Am. Chem. Soc.* **2006**, *128*, 3148. (b) Watanabe, Y.; Aoki, S.; Tanabe, D.; Setiawan, A.; Kobayashi, M. *Tetrahedron* **2007**, *63*, 4074. (c) Aoki, S.; Watanabe, Y.; Tanabe, D.; Setiawan, A.; Arai, M.; Kobayashi, M. *Tetrahedron Lett.* **2007**, *48*, 4485. (d) Aoki, S.; Watanabe, Y.; Tanabe, D.; Arai, M.; Suna, H.; Miyamoto, K.; Tsujibo, H.; Tsujikawa, K.; Yamamoto, H.; Kobayashi, M. *Bioorg. Med. Chem.* **2007**, *15*, 6758.

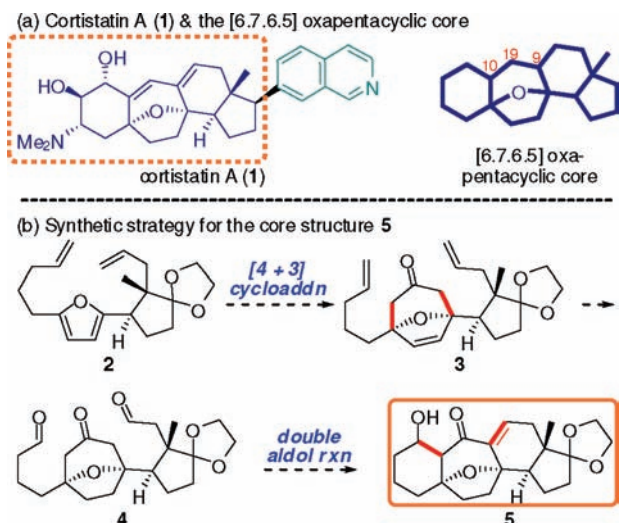


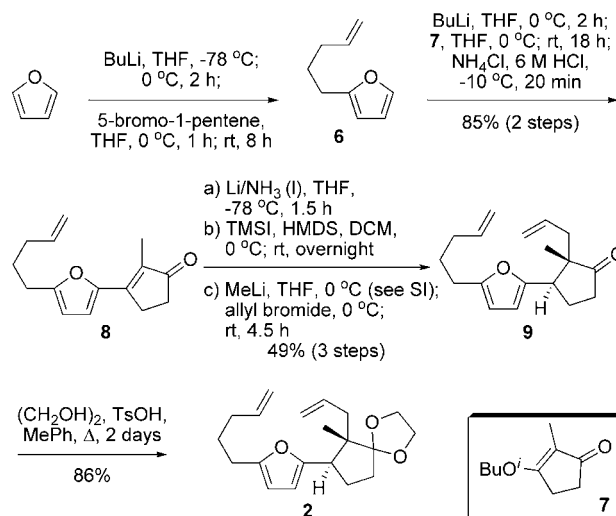
Figure 1. Structural features of cortistatin A and synthetic strategy for core structure **5**.

of **1** have been conducted by a number of research groups.⁹ Moreover, certain novel cortistatin analogues have been synthesized and their biological activities evaluated.^{4b,10}

Furan-involving [4 + 2] cycloaddition was utilized in the core construction studies by both Yang⁹ⁱ and Magnus.^{9j} We envisaged that a core structure of cortistatin A (**1**) such as **5** (Figure 1b) may be rapidly assembled via a furan–oxyallyl [4 + 3] cycloaddition^{9d} (**2** → **3**) and double-intramolecular aldol reactions (**4** → **5**).

As outlined in Scheme 1, furan was alkylated to afford the known intermediate **6**,¹¹ which was lithiated at C-5 and reacted with ketone **7**¹² followed by hydrolysis to give enone **8** in 85% yield over two steps. Stereoselective reductive α -allylation¹³ of the cyclopentenone moiety in **8** led effectively to ketone **9** with a quaternary carbon center (49%) via a three-step sequence consisting of (i) Birch reduction (Li/NH₃ (l), THF, -78 °C), (ii) thermodynamic enol silyl ether formation (TMSI, HMDS, DCM, 0 °C to rt), and (iii)

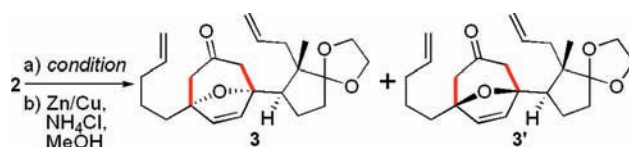
Scheme 1. Synthesis of Furan Derivative **2**



regiospecific enolate generation and subsequent alkylation (MeLi, THF, 0 °C; allyl bromide, THF, rt). Treatment of **9** with ethylene glycol in the presence of TsOH in toluene at reflux furnished acetal **2** in 86% yield.

With 2,5-disubstituted furan derivative **2** in hand, the [4 + 3] cycloaddition¹⁴ was extensively investigated under a series of conditions (Scheme 2). Although the cycloaddition

Scheme 2. Intermolecular [4 + 3] Cycloaddition of **2**



entry	condition	result
1	TBA (2.5 equiv), Et ₂ Zn (2 equiv), MePh, 2 d	3 (12%) + 3' (15%)
2	TCA (1.5 equiv), TEA (3 equiv), TFE, 4 d	most of 2 unreacted
3	TCA (1.5 equiv), TEA (3 equiv), HFIP	3 43% + 3' (53%)
4	TCA (3 equiv), TEA (4 equiv), HFIP	3 (42%) + 3' (45%)
5	TCA (5 equiv), TEA (10 equiv), HFIP	3 (46%) + 3' (47%)

reaction (under improved Föhlisch's conditions¹⁵) seemed to generate slightly less of the desired cycloadduct **3** than the other isomer **3'** after reductive dehalogenation, the combined yield of both stereoisomers reached as high as 96% (entry 3). The highest yield of **3** (46%) was obtained when

(14) For reviews on oxyallyl [4 + 3] cycladditions, see: (a) Noyori, R.; Hayakawa, Y. *Org. React.* **1983**, *29*, 163. (b) Hoffmann, H. M. R. *Angew. Chem., Int. Ed. Engl.* **1984**, *23*, 1. (c) Rigby, J. H.; Pigge, F. C. *Org. React.* **1997**, *51*, 351. (d) Hartung, I. V.; Hoffmann, H. M. R. *Angew. Chem., Int. Ed.* **2004**, *43*, 1934. For recent examples, see: (e) Lee, J. C.; Cha, J. K. *J. Am. Chem. Soc.* **2001**, *123*, 3243, and references cited therein. (f) See also ref 11.

(15) Sendelbach, S.; Schwetzer-Raschke, R.; Radl, A.; Kaiser, R.; Hendle, G. H.; Korfant, H.; Reiner, S.; Föhlisch, B. *J. Org. Chem.* **1999**, *64*, 3398, and references cited therein.

(8) Flyer, A. N.; Si, C.; Myers, A. G. *Nature Chem.* **2010**, *2*, 886 (total synthesis).

(9) (a) Simmons, E. M.; Hardin, A. R.; Guo, X.; Sarpong, R. *Angew. Chem., Int. Ed.* **2008**, *47*, 6650. (b) Yamashita, S.; Iso, K.; Hiramata, M. *Org. Lett.* **2008**, *10*, 3413. (c) Kurti, L.; Czako, B.; Corey, E. J. *Org. Lett.* **2008**, *10*, 5247. (d) Craft, D. T.; Gung, B. W. *Tetrahedron Lett.* **2008**, *49*, 5931. (e) Kotoku, N.; Sumii, Y.; Hayashi, T.; Kobayashi, M. *Tetrahedron Lett.* **2008**, *49*, 7078. (f) Dai, M. J.; Danishefsky, S. J. *Tetrahedron Lett.* **2008**, *49*, 6610. (g) Dai, M. J.; Wang, Z.; Danishefsky, S. J. *Tetrahedron Lett.* **2008**, *49*, 6613. (h) Dai, M. J.; Danishefsky, S. J. *Heterocycles* **2009**, *77*, 157. (i) Liu, L.; Gao, Y.; Che, C.; Wu, N.; Wang, D. Z.; Li, C. C.; Yang, Z. *Chem. Commun.* **2009**, 662. (j) Magnus, P.; Littich, R. *Org. Lett.* **2009**, *11*, 3938. (k) Frie, J. L.; Jeffrey, C. S.; Sorensen, E. J. *Org. Lett.* **2009**, *11*, 5394.

(10) (a) Czako, B.; Kurti, L.; Mammoto, A.; Ingber, D. E.; Corey, E. J. *J. Am. Chem. Soc.* **2009**, *131*, 9014. (b) Cee, V. J.; Chen, D. Y. K.; Lee, M. R.; Nicolaou, K. C. *Angew. Chem., Int. Ed.* **2009**, *48*, 8952. (c) Shi, J.; Shigehisa, H.; Guerrero, C. A.; Shenvi, R. A.; Li, C.-C.; Baran, P. S. *Angew. Chem., Int. Ed.* **2009**, *48*, 4328.

(11) Chung, W. K.; Lam, S. K.; Lo, B.; Liu, L. L.; Wong, W. T.; Chiu, P. *J. Am. Chem. Soc.* **2009**, *131*, 4556.

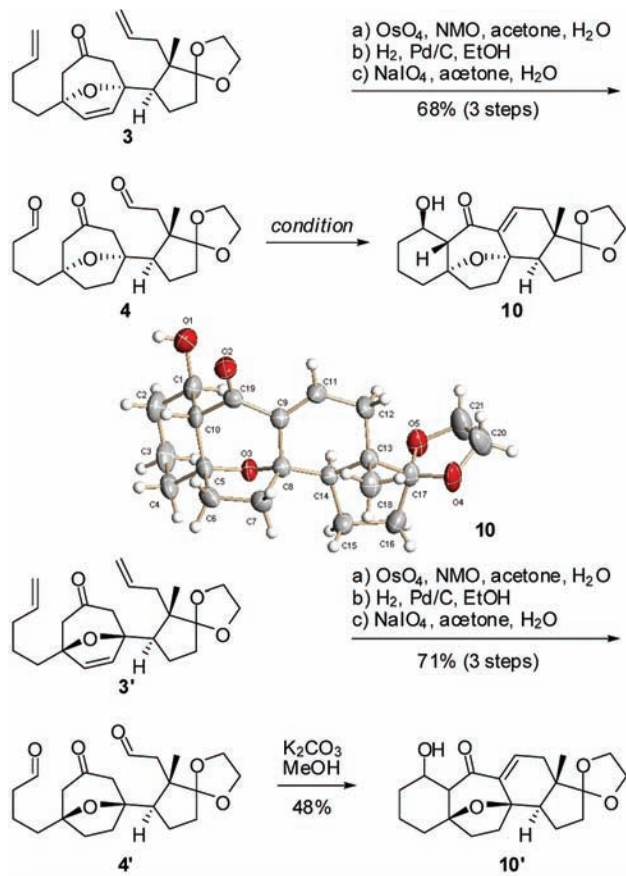
(12) Funk, R. L.; Vollhardt, K. P. C. *Synthesis* **1980**, 118.

(13) Van Royen, L. A.; Mijngheer, R.; De Clercq, P. *J. Bull. Soc. Chim. Belges.* **1984**, *93*, 1019.

2 was treated with 1,1,3-trichloroacetone (5 equiv) and triethylamine (10 equiv) in 1,1,1,3,3,3-hexafluoro-2-propanol (entry 5).

Both **3** and **3'** were further converted into more advanced intermediates with an attractive [6.7.6.5] tetracarboxylic framework (Scheme 3). Upon terminal alkene dihydroxylation,

Scheme 3. Synthesis of Core Structures **10** and **10'**



entry	condition (4 → 10)	result
1	4 (0.02 M), L-proline, MeCN, 3 d	NR
2	4 (0.02 M), K ₂ CO ₃ (5 equiv), dioxane, 3 d	NR
3	4 (0.05 M), K ₂ CO ₃ (5 equiv), MeOH, 2 h	10 (62%) + 11 (10%)
4	4 (0.05 M), K ₂ CO ₃ (5 equiv), MeOH, 0.5 h	10 (71%) + 11 (6%)
5	4 (0.02 M), K ₂ CO ₃ (5 equiv), MeOH, 0.5 h	10 (82%)

Compound **11** is a dienone ($\Delta^{1,10,9,11}$) derived from **10** via dehydration.

tion, internal alkene hydrogenation, and vicinal diol cleavage,¹⁶ ketodialdehyde **4** was formed from triene **3** in 68% overall yield. Condensation parameters such as the substrate concentration, the base, the solvent, and the reaction time

were carefully scrutinized for the double-aldol reaction (see the table at the bottom of Scheme 3). The best yield (82%) for tetracarboxycle **10** was obtained when **4** (0.02 M) was exposed to K₂CO₃ (5 equiv) in MeOH for 30 min (entry 5). In this case, the double dehydration product (dienone **11**, $\Delta^{1,10,9,11}$) was not observed, which turned out to be a superior result since hydroxy enone **10** may be a better substrate for further structural manipulations than dienone **11**. The structure of **10** was unambiguously confirmed by X-ray crystallographic analysis. Analogously, triene **3'** was transformed into ketodialdehyde **4'** in three steps, while the overall yield was slightly higher (71%) compared to that of **4**. Double aldol reaction of **4'** in the presence of K₂CO₃ in MeOH for 2 days resulted in double cyclization to furnish **10'** (48%) as a single isomer, although the relative configuration for the newly formed stereogenic centers in **10'** remains uncertain at this stage.¹⁷

In summary, we have described a short synthetic approach for constructing the oxapentacyclic framework of cortistatin A featuring a furan–oxyallyl [4 + 3] cycloaddition^{9d} and double-intramolecular aldol reactions. Core structure **10**, as well as isomeric **10'**, was obtained in 10 steps from the known furan derivative **6** or in 11 steps from furan. The current strategy has the full potential to lead to a total synthesis due to the proper arrangement of the desired functionalities in our target molecule that are amenable for further elaborations.

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Supporting Information Available: Experimental procedures and analytical data (CIF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(16) Uoto, K.; Takenoshita, H.; Yoshino, T.; Hirota, Y.; Ando, S.; Mitsui, I.; Terasawa, H.; Soga, T. *Chem. Pharm. Bull.* **1998**, *46*, 770.

(17) The structure of **10'** can presumably be represented by:

